

**I B. Tech II Semester Regular Examinations, September- 2021**  
**ENGINEERING CHEMISTRY**  
(Comm. to CE, ME, Agri E)

Time: 3 hours

Max. Marks: 70

**Answer any five Questions one Question from Each Unit**  
**All Questions Carry Equal Marks**

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**UNIT-I**

- 1 a) Define polymerization. Explain emulsion polymerization and mention its advantages. (7M)  
b) Explain preparation and application of BUNA-S and polycarbonates. (7M)

Or

- 2 a) Discuss the recycling of e-plastic waste and mention its advantages. (7M)  
b) When does a polymer conduct electricity? Explain n-doped conducting polymers. (7M)

**UNIT-II**

- 3 a) What is electrode potential? Explain determination of single electrode potential. (7M)  
b) Discuss electrochemical theory of corrosion. (7M)

Or

- 4 a) Define battery. Explain working principle involved in zinc-air cell with neat sketch. (7M)  
b) Explain (i) electroplating (ii) electroless plating of nickel. (7M)

**UNIT-III**

- 5 a) Define nanomaterials. Explain preparation of metal oxides by sol-gel method. (7M)  
b) Explain manufacturing of cement. (7M)

Or

- 6 a) Discuss the instrumentation and applications of TGA (7M)  
b) Explain thin film and thick film lubrication. (7M)

**UNIT-IV**

- 7 a) Define calorific value. Differentiate the higher and lower calorific values. (7M)  
b) Explain flue gas analysis by Orsat apparatus. (7M)

Or

- 8 a) A coal has the following composition by weight: C = 90%, O = 3 %, S = 0.5%, N= 0.5%, H = 4% and remaining ash. Calculate the net and gross calorific value of coal. Latent heat of steam = 587 cal/g. (7M)  
b) Explain synthesis of petrol by Bergius process. (7M)

**UNIT-V**

- 9 a) What is hardness of water? Mention its units and types of hardness. (7M)  
b) Write notes on (i) break-point chlorination (ii) reverse osmosis (7M)

Or

- 10 a) Explain softening of hard water by cation and anion exchange resins. (7M)  
b) Discuss in detail about boiler corrosion. (7M)

